The Acylation of bz-Hydroxy- and bz-Methoxy-2,3-dimethylbenzofurans and the Synthesis of Furobenzopyranone Derivatives¹⁾

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The action of acetic acid, phenylacetic acid, or β -phenylpropionic acid and polyphosphoric acid on 5-, 7-, and 6-hydroxy-2,3-dimethylbenzofurans afforded 6-acyl-5-hydroxy-, 6-acyl-7-hydroxy-, and 5-acyl-6-hydroxy-benzofurans, some of which were also obtained by the Friedel-Crafts acylation of the corresponding methoxy-benzofurans. The hydroxyketones thus obtained were converted to dimethylfuro derivatives of 4-hydroxycoumarins and isoflavones, and 3-benzyl-4-hydroxyfurocoumarins were also prepared by the thermal condensation of the hydroxybenzofurans and diethyl benzylmalonate.

The formylation and acetylation of bz-methoxy-2,3-dimethylbenzofurans have been reported by Royer $et\ al.,^2$) while the acetylation and phenylacetylation of 4- and 6-methoxy- and -hydroxybenzofurans have been reported by one of the present authors.³⁾ The acyl compounds obtained have been converted to dimethylfuro derivatives of coumarins, chromones, flavones, and isoflavones.²⁻⁴⁾

In connection with this, the acetylation, phenylacetylation, and β -phenylpropionylation of 5-, 7-, and 6-hydroxy- (3, 9, 15) and -methoxy-2,3-dimethylbenzofurans (2, 8, and 14) were carried out in the present experiments; also, dimethylfuro derivatives of 4-hydroxycoumarins (17, 18, and 19) and of isoflavones (20, 21, 22, and 23) were prepared from the acyl compounds in order to test the pharmacological activities.

The starting compounds, bz-methoxy-2,3-dimethylbenzofurans (2, 8, and 14), were prepared in good yields in the present experiments by the cyclodehydration of aryloxybutanones (1, 7, and 12) with polyphosphoric acid. The reagent seemed to be excellent in the case of 3-(m-methoxyphenoxy)-2-butanone⁵⁾ (12), as the 6-methoxybenzofuran (14) was obtained in the pure state.⁶⁾ For the preparation of the 6- or 4-hydroxybenzofuran, the cyclization of 3-(m-acetoxyphenoxy)-2-butanone (13) by sulfuric acid was at-

tempted in order to obtain the 6-hydroxybenzofuran (15) in a low yield.

The action of acetic acid, phenylacetic acid, or β -phenylpropionic acid on the 5- and 7-hydroxy-2,3-dimethylbenzofurans⁷) (3 and 9) in the presence of polyphosphoric acid afforded the corresponding 6-acyl compounds (5a, 5b, 5c, 11a, 11b, and 11c), while the analogous β -phenylpropionylation of the 6-hydroxy-benzofuran⁸) (15) afforded the 5-(β -phenylpropionyl) compound (16) (Charts 1—3).

The 6-acyl-5-hydroxy compounds (5a and 5b) were also obtained by the Friedel-Crafts acylation of the methoxybenzofuran (2) in carbon disulfide at the reflux temperature, while the β -phenylpropionylation under those conditions afforded the 6-acyl-5-methoxybenzofuran (4c), which was then dimethylated to 5c by heating it with aluminum chloride in nitrobenzene (Chart 1). Royer et al.2) have obtained the 6-acetyl-5-methoxybenzofuran (4a) by the Friedel-Crafts acetylation of the 5-methoxybenzofuran (2) at room temperature and then changed it to the hydroxyketone (5a) by demethylation. The Friedel-Crafts β phenylpropionylation of the 7-methoxybenzofuran (8) afforded mainly the 4-acyl-7-methoxybenzofuran (10c), accompanied by a small amount of the 6-acyl-7hydroxy compound (11c) (Chart 2) analogously

Chart 1

1) The major part of this work was presented at the 23rd Annual Meeting of the Chemical Society of Japan, Tokyo, April, 1970.

2) R. Royer, E. Bisagni, A.-M. Laval-Jeantet, and J.-P. Marquet, Bull. Soc. Chem. Fr., 1965, 2607.

3) Y. Kawase, M. Nanbu, and F. Miyoshi, This Bulletin, 41, 2676 (1968).

4) Y. Kawase, M. Nanbu, F. Miyoshi, and H. Kawamura, *ibid.*, **41**, 2683 (1968).

5) Preliminary report: Y. Kawase, Chem. Ind. (London),

1970, 687

6) The reported cyclization by sulfuric acid or other reagents afforded a mixture of the 6- and 4-methoxybenzofurans (Ref. 7 and 8); their ratio was determined to be 97:3 in the case of sulfuric acid by estimating the areas of the peaks corresponding to each of the 3-methyl protons in the NMR spectrum (Ref. 5).

7) R. Royer, E. Bisagni, C. Hudry, A. Cheutin, and M.-L. Desvoye, *Bull. Soc. Chim. Fr.*, **1963**, 1003.

8) E. Bisagni and R. Royer, ibid., 1962, 925.

to what has been reported in the case of the acetylation.²⁾ The analogous β - phenylpropionylation of the 6-methoxybenzofuran (14) aforded also the 5-acyl-6-hydroxybenzofuran (16c), similarly as has been reported in the cases of the acetylation and phenylacetylation.³⁾ The Friedel-Crafts acetylation of 2 with a double amount of the reagents afforded the diacetyl compound (6), analogously to the cases of 5-methylbenzofuran⁹⁾ and 7-methoxybenzofuran.²⁾

The positions of the acyl groups in the acylation product were determined by NMR spectroscopy; the methoxyketone ($\mathbf{4c}$) and hydroxyketones ($\mathbf{5a}$, $\mathbf{5b}$, $\mathbf{5c}$, and $\mathbf{16c}$) have two singlets corresponding to two aromatic para protons, while the methoxyketone ($\mathbf{10c}$) and hydroxyketones ($\mathbf{11a}$, $\mathbf{11b}$, and $\mathbf{11c}$) have two doublets of J=8 Hz corresponding to two

aromatic ortho protons (Table 3).

The hydroxyketones (5a, 5b, 5c, 11a, 11b, 11c, and 16c) thus obtained were converted to dimethylfuro derivatives of 4-hydroxycoumarins (17a, 17b, 17c, 18a, 18b, 18c, and 19c) by the action of diethyl carbonate and sodium, 4) and to dimethylfuro derivatives of isoflavones (20, 21, 22, and 23) by the action of ethyl orthoformate and piperidine in pyridine or acetic anhydride and sodium acetate. 3) The furocoumarins (17c, 18c, and 19c) were also prepared by the thermal condensation of the hydroxybenzofurans (3, 9, and 15) with diethyl benzylmalonate, analogously as has been reported in the cases of 3-phenylfurocoumarins 10) (17b, 18b, and 19b) (Chart 4).

Experimental

All the melting points and boiling points are uncorrected. The IR spectra were measured as KBr disks on a Hitachi EPI-S spectrophotometer, the UV spectra were measured on a Hitachi 139 spectrophotometer, the NMR spectra were measured on a JEOL JNM-C-60H (60 MHz) spectrometer, and the mass spectra were measured on a JEOL JMS-OIS spectrometer. The detailed data are summarized in Tables 1—4.

The Preparation of Methoxybenzofurans. A mixture of $3-(p\text{-methoxyphenoxy})-2\text{-butanone}^{7)}$ (1) (3 g) and polyphosphoric acid (n=2.5, 60 g) was heated at 100°C for 1 hr, after which the cooled mixture was poured into ice water. The mixture was extracted with ether, and the residual product from the ethereal solution was distilled to give methoxybenzofuran (2), bp $143-147^{\circ}\text{C}/21 \text{ mmHg}$ (lit,7) bp $133-134^{\circ}\text{C}/11 \text{ mmHg}$); 1.5 g (55%). Similarly, **8** and

⁹⁾ R. Royer, Y. Kawase, M. Hubert-Habart, L. René, and A. Cheutin, Bull. Soc. Chim. Fr., 1966, 211.

¹⁰⁾ J.-P. Lechartier, P. Demerseman, A. Cheutin, and R. Royer, *ibid.*, **1966**, 1716.

TABLE 1. SUMMARIZED DATA OF REACTIONS

Starting compd.	Reagent	Prod.	$egin{array}{cc} ext{Mp} \ ^{\circ} ext{C(solv}^{a)} \ ext{or bp} \ ^{\circ} ext{C/mmHg} \end{array}$	Yield %
	Preparati	ion of Aryloxybut	anone	
b)	Cl-butanone	13	145—150/7	72.5
	Preparation of Mo	ethoxy- and Hydr	oxybenzofurans	
1	PPA ^{c)}	2	143—147/21 ^{d)}	55
7	PPA	8	143—147/25°)	59
12	PPA	14	141—144/21 ^{f)}	81
13	H_2SO_4	15	105-107 (Et) g)	17
	Preparation of 1	Methoxy- and Hy	droxyketones	
2	PhCH ₂ CH ₂ COCl - AlCl ₃	4c	105—108 (Et)	60
2	AcCl – AlCl ₃	5 a	140—141 (Et) h)	81
3	AcOH - PPA	5 a	140—141 (Et) h)	39
2	PhCH ₂ COCl – AlCl ₃	5 b	134—135 (Et)	58
3	PhCH ₂ CO ₂ H - PPA	5 b	134—135 (Et)	15
3	$PhCH_2CH_2CO_2H - PPA$	5 c	139—140.5 (Et)	31.5
4c	AlCl ₃	5 c	139—140.5 (Et)	34
2	AcCl - AlCl ₃	6	150.5—151 (Et)	67
8	PhCH ₂ CH ₂ COCl - AlCl ₃	(10c i)	64—66 (Me)	42
	2 2	(11c	$98-98.5({ m Me})$	6
9	AcOH - PPA	11a	$127-129.5 (Et)^{j}$	41
9	$PhCH_2CO_2H - PPA$	11b	159.5—161.5 (Et)	48.5
9	$PhCH_2CH_2CO_2H - PPA$	11c	98-98.5(Et)	36
14	PhCH ₂ CH ₂ COCl – AlCl ₃	16c ^k)	100—101 (Et)	2
15	$PhCH_2CH_2CO_2H - PPA$	16c	100—101 (Et)	8
	Preparat	ion of Furocoum	arins	
5 a	$CO(OEt)_2$ – Na	17a	308 (Et)	67.5
5b	$CO(OEt)_2$ – Na	17b	284—285 (Et) 1)	72.5
3	$PhCH_2CH(CO_2Et)_2$	17c	285—286 (Et)	55
5c	$CO(OEt)_2$ – Na	17c	285—286 (Et)	30
lla	$CO(OEt)_2$ – Na	18a	310 (Et)	29.5
11b	$CO(OEt)_2$ – Na	18 b	283—284(Et) ^{m)}	52
9	$PhCH_2CH(CO_2Et)_2$	18c	283.5-285 (Et)	42
11c	$CO(OEt)_2$ – Na	18c	283.5 - 285 (Et)	21
15	$PhCH_2CH(CO_2Et)_2$	19c	$269-270 (\mathrm{Et})$	8
16c	$CO(OEt)_2$ – Na	19c	$269-270 (\mathrm{Et})$	74
	Preparat	ion of Furoisoflav	ones	
5 a	CH(OEt) ₃	20	233—234 (Et)	97
5a	Ac ₂ O - AcONa	21	150.5—151.5 (Et)	28
11a	$CH(OEt)_3$	22	216—217 (Et)	85
11a	$Ac_2O - AcONa$	23	198—201 (Et)	61

a) Et: ethanol, Me: methanol. b) m-Acetoxyphenol. c) Polyphosphoric acid. d) Lit. bp 133—134°C/11 mmHg (Ref. 7). e) Lit. bp 136°C/12 mmHg, mp 38.5°C (Ref. 7). f) Lit. bp 128—129°C/9 mmHg (Ref. 7). g) Lit. mp 107.5—108°C (Ref. 8). h) Lit. mp 141°C (Ref. 2). i) The crude product was crystallized from methanol to give 11c, and 10c was obtained from the mother solution. j) Lit. mp 130°C (Ref. 2). k) The crude product was distilled to give fraction a, bp 134—140°C/20 mmHg, and fraction b, bp 150—170°C/20 mmHg. The fraction b was crystallized to give 16c, and the fraction a was purified by chromatography on silica gel, with cyclohexane and then benzene as the solvent, to give the recovery 14 (total 39%) and indanone (total 68%). l) Lit. mp 187°C (Ref. 10). m) Lit. mp 287°C (Ref. 10).

14 were also prepared by this procedure.

The Preparation of Hydroxybenzofurans. Hydroxybenzofurans (3, 9, and 15) were prepared by the demethylation of the methoxybenzofurans (2, 8, and 14) by following the methods of Refs. 7 and 8. Compound 15 was also prepared by the cyclization of 13 as follows:

The Preparation of 3-(m-Acetoxyphenoxy)-2-butanone (13). A mixture of m-acetoxyphenol (28 g), 3-chloro-2-butanone (20.5 g), acetone (150 ml), and anhydrous potassiun carbonate (70 g) was refluxed for 8 hr; the mixture was treated

with water and then extracted with ether. The residual product from the ethereal solution was distilled to give 13, bp $145-150^{\circ}$ C/7 mmHg; 29.6 g (72.5%).

The Cyclization of 13. Concentrated sulfuric acid (20 ml) was stirred into 13 (13.5 g) with cooling below 15°C, after which the mixture was kept at 15°C for 1 hr. The resulting mixture was poured into ice water and extracted with ether. The residual product from the ethereal solution was crystallized from cyclohexane to give 15, mp 105—107°C (lit,7) mp 107.5—108°C); 1.7 g (17%).

Table 2. The IR spectra and analysis of new compounds

Compd. v_{co}^{KBr}	KBr	Formula	For	Found		Calcd	
	v_{co}^{ab}		G%	H%	G%	H%	$m/e({f M}^+$
			Aryloxybu	tanone			
13	{1775 \1729	$\mathrm{C_{12}H_{14}O_4}$	64.82	6.45	64.85	6.35	
		Me	thoxy- and Hy	droxy-ketones	6		
4	1664	${ m C_{20}H_{20}O_{3}}$	77.68	6.49	77.90	6.54	308
5 b	1655	$\mathrm{C_{18}H_{16}O_3}$	77.00	5.70	77.12	5.75	280
5c	1658	$\mathrm{C_{19}H_{18}O_3}$	77.51	6.12	77.53	6.16	294
6	{1671 {1626	$\mathrm{C_{14}H_{14}O_4}$	68.21	5.44	68.28	5.73	246
10c	1674	${ m C_{20}H_{20}O_{3}}$	77.85	6.51	77.90	6.54	308
11b	1664	$\mathrm{C_{18}H_{16}O_3}$	76.96	5.74	77.12	5.75	280
11c	1658	${ m C_{19}H_{18}O_3}$	77.75	6.30	77.53	6.16	294
16c	1632	$\mathrm{C_{19}H_{18}O_3}$	77.69	6.21	77.53	6.16	294
			Furocoum	arins			
17a	1692	$C_{13}H_{10}O_{4}$	68.02	4.39	67.82	4.38	230
17c	1670	$\mathrm{C_{20}H_{16}O_4}$	75.10	4.80	74.99	5.03	320
18a	1678	$C_{13}H_{10}O_{4}$	67.51	4.38	67.82	4.38	230
18c	1672	$C_{20}H_{16}O_{4}$	75.21	4.83	74.99	5.03	320
19c	1650	$\mathrm{C_{20}H_{16}O_4}$	74.95	5.12	74.99	5.03	320
			Furoisofla	vones			
20	1625	$\mathrm{C_{19}H_{14}O_3}$	78.60	4.82	78.60	4.85	290
21	1637	$\mathrm{C_{20}H_{16}O_{3}}$	78.79	5.19	78.93	5.30	304
22	1637	$\mathrm{C_{19}H_{14}O_3}$	78.56	4.80	78.60	4.85	290
23	1635	$\mathrm{C_{20}H_{16}O_3}$	78.73	5.35	78.93	5.30	304

TABLE 3. THE NMR SPECTRA®)

Compd.	Ph-H of I furan		Ph-H	2-Me	3-Ме	Compd.	Ph-H of benzo- furan ^{b)}		Ph-H	2-Me	3-Me
Methoxybenzofurans					11a ^{d)}	5 H	4 H				
2				2.32	2.08		7.49(d)	6.88(d)		2.41	2.12
8				2.35	2.10			0. II			
14				2.24	2.00		J =	8 Hz			
c)				2.24	2.24	11b	5H	4H			
- /	Mark	ad U-		tomac			7.52(d)	6.73(d)	7.24	2.43	2.12
Methoxy- and Hydroxyketones											
4c	7H	4H	7 10	0.00	0.05		J=	3 Hz			
_	7.60	6.67	7.12	2.30	2.05	11c	5H	4H			
5a	7 H	4H		2.36	2.08		7.39(d)	6.70(d)	7.17	2.38	2.08
	7.54	6.76		2.30	2.00			``			
5b	7H 7.61	4H 6.75	7.25	2.32	2.04		J=8 Hz				
= 4)	· -		7.23	4.54	2.01	16-	•	7 H			
$\mathbf{5c}^{ ext{d}}$	7H	4H 6.85	7.26	2.34	2.06	16a	4H 7,40	6.68		2.28	2.03
C4)	7.60	0.03	7.20	2.37	1.97					4.40	4.03
6 d)	7.70	CII		4.37	1.97	16b ^{d)}	4H	7 H			
10c	5H	6H	7 14	0.25	2.10		7.82	6.91	7.34	2.33	2.11
	7.31 (d)	6.48(d)	7.14	2.35	4.10	16c	4H	7H			
J=8 Hz					_00	7.48	6.75	7.18	2.28	2.05	

a) δ -Values in CCl₄ (about 5% solution), with TMS as the internal standard. b) d: Doublet. c) 2,3-Dimethyl-4-methoxybenzofuran. d) In CDCl₃.

The Acylation of the Hydroxybenzofuran with Carboxlyic Acid and Polyphosphoric Acid (PPA). A mixture of $\mathbf{3}^{7}$ (2.4 g). acetic acid (1 g), and PPA (n=1.5, 45 g) was heated at 100° C for 1 hr. The cooled mixture was poured into ice water and then extracted with chloroform. The residual product from the chloroform solution was crystallized from ethanol to give $5\mathbf{a}$, mp $140-141^{\circ}$ C (lit, 20) mp 141° C); 1.3 g (39%).

Similarly, the acetyl, phenylacetyl, and β -phenylpropionyl compounds (11a, 5b, 11b, 5c, 11c, and 16c) were also prepared by this procedure.

The Friedel-Crafts Acylation of the Methoxybenzofurans. Powdered aluminum chloride (9.7 g) was added, with stirring and cooling, to a solution of 2^{7} (5 g) and acetyl chloride (2.5 g) in carbon disulfide (50 ml); the mixture was stirred

TABLE 4. THE UV SPECTRA

Compd. $\lambda_{\max}^{\text{EtoH}} \ \text{m} \mu^{a} \ (\log \epsilon)$						
Methoxybenzofurans						
2	214 (4.33), 255 (4.02), 292.5 (3.68), 301 (3.62)					
8	217(4.41), 251(4.09), 285(3.01)					
14	217 (4.24), 249 (4.06), 256 (4.06), 292 (3.72)					
b)	$215.5(4.34)$, $258(4.08)$, $276^{s}(3.55)$, $287(3.45)$					
Hydroxybenzofurans						
3	216(5.36), 255(5.13), 295(4.73)					
9	216 (4.36), 250 (4.12), 255.5 (4.12), 286 (3.16)					
15	216(5.36), 248(5.09), 255(5.08), 293(4.76)					
c)	218 (5.35), 252 (5.03), 281 (4.44), 289 (4.41)					
Hydroxyketones						
5a	230(4.20), 316(4.30)					
5 b	213.5(4.18), 231(4.22), 318(4.33)					
11a	239 (4.33), 296 (4.16), 336 (3.96)					
11b	211 (4.38), 239 (4.32), 301 (4.17), 341 (4.01)					
16a	240 (4.56), 267 (3.97), 360 (3.55)					
16b	240(4.55), 360(3.57)					
\mathbf{d})	243(4.49), 265.5(3.98), 280.5(3.82), 345(3.51)					
e)	211.5(4.16), $244(4.50)$, $253(4.47)$, $283(3.95)$,					
	348 (3.56)					
	Furocoumarins					
17a	214 (4.50), 323 (4.49), 338 (4.28)					
17b	$216(4.55)$, $332(4.42)$, $342^{s}(4.40)$					
17c	216(4.52), 328.5(4.33), 340(4.25)					
18a	212(4.40), $239.5(4.45)$, $245.5(4.44)$,					
	$265^{s}(3.98)$, $285^{s}(3.92)$, $296(4.04)$, $323.5(4.06)$					
18Ь	$215.5(4.45)$, $236.5(4.38)$, $250^{s}(4.33)$,					
	$302^{s}(4.00)$, $332(4.22)$					
18c	217(4.49), $249(4.36)$, $300.5(4.03)$, $326(4.11)$					
19c	216(4.47), $246(4.46)$, $253(4.40)$, $301(4.14)$,					
	320 (4.11), 331s (4.08)					
Furoisoflavones						
20	253 (4.43), 315 (4.25)					
21	240 (4.43), 312 (4.28)					
22	261 (4.59), 311 (4.03)					
23	257 (4.57), 309 (4.08)					

a) s: Shoulder. b) 2,3-Dimethyl-4-methoxybenzofuran. c) 2,3-Dimethyl-4-hydroxybenzofuran. d) 2,3-Dimethyl-4-hydroxy-5-benzofuranyl methyl ketone. e) 2,3-Dimethyl-4-hydroxy-5-benzofuranyl benzyl ketone.

at room temperature for 2 hr and then at the reflux temperature for 1 hr. The cooled mixture was poured into ice water containing hydrochloric acid and extracted with chloroform. The chloroform solution was washed with an aqueous sodium hydroxide solution, and the residual product from the chloroform solution was crystallized from ethanol to give 5a, mp $140-141^{\circ}$ C, identical with the sample described above. Yield, 4.7 g (81%). Similarly, 5b was obtained by the action of phenylacetyl chloride, while the methoxyketone (4c) was obtained in the case of β -phenyl-

propionyl chloride. The β -phenylpropionylation of **8** afforded a small amount of **11c** and a fairly large amount of the methoxyketone (**10c**), while that of **14** afforded a small amount of **16c** and fairly large amounts of the recovered **14** and indanone. The diacetylation of **2** afforded the hydroxydiketone (**6**).

The Denethylation of the Methoxyketone (4c). A mixture of 4c (1.4 g), nitrobenzene (40 ml), and powdered aluminum chloride (0.9 g) was heated at 100°C for 1 hr. The cooled mixture was poured into dilute hydrochloric acid, and the nitrobenzene was removed by steam distillation. The residual precipitates obtained were crystallized from ethanol to give 5c, mp 139—140.5°C, identical with the sample described above. Yield, 0.46 g (34%).

Preparation of Furocoumarins. a) From Hydroxyketones: Small pieces of sodium (1.1 g) was added to a mixture of **5a** (1.1 g) and diethyl carbonate (40 ml), after which the mixture was heated at 110-120°C for 30 min. The cooled mixture was treated with a small amount of methanol and with water, and was then extracted with ether. The aqueous layer was acidified, and the crystalline precipitates thus formed were recrystallized from ethanol to give 6H-8-hydroxy-2,3-dimethylfuro[2,3-g][1]benzopyran-6one (17a), mp 308°C (dec.); 0.8 g (67.5%). Similarly, 7phenyl and 7-benzyl derivatives (17b and 17c) of 17a, 8H-6-hydroxy-2,3-dimethylfuro[3,2-h][1]benzopyran-8-one (18a) and its derivatives (18b and 18c), and 7H-6-benzyl-5-hydroxy-2,3-dimethylfuro[3,2-g][1]benzopyran-7-one (19c) were also prepared by this procedure.

b) From the Hydroxybenzofurans: A mixture of 3 (0.5 g) and diethyl benzylmalonate (2.5 g) was refluxed for 8 hr. The cooled mixture was washed with ether, and the residual solid was crystallized from ethanol to give 17c, mp 285—286°C, identical with the sample described above. Yield, 0.54 g (55%). Similarly, 18c and 19c were also prepared by this procedure.

The Preparation of Furoisoflavones. a) By the Action of Ethyl Orthoformate: A mixture of the hydroxyketone **5b** (0.4 g), ethyl orthoformate (4 g), piperidine (1 drop), and pyridine (16 ml) was refluxed for 8 hr, after which the cooled mixture was poured into dilute hydrochloric acid to form crystalline precipitates, which were then recrystallized from ethanol to give 8H-2,3-dimethyl-7-phenylfuro[2,3-g][1]benzopyran-8-one (**20**), mp 233—234°C; 0.4 g (97%). Similarly, 6H-2,3-dimethyl-7-phenylfuro[3,2-h][1]benzopyran-6-one (**22**) was also prepared by this procedure.

b) By the Action of Ac_2O -AcONa: A mixture of **5b** (0.5 g), anhydrous sodium acetate (2 g), and acetic anhydride (10 ml) was refluxed for 12 hr, the cooled mixture was then poured into water to form crystalline precipitates, which were recrystallized from ethanol to give the 6-methylderivative (21) of 20, mp 150.5—151.5°C; 0.15 g (28%). Similarly, the 8-methyl derivative (23) of 22 was also prepared by this procedure.

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